



Entrance Exam

PHYSICS

Name: _____

Circle the correct answer.

1. A body of 3 kg mass and 5 g/cm^3 density has the volume:
 - a) 600 litres
 - b) 6 dm^3
 - c) 0.6 litres
2. When a body of 20 g mass reaches velocity of 200 m/s, its kinetic energy becomes:
 - a) 400 J
 - b) 800 J
 - c) 400 W
3. Temperature of -13°C equals:
 - a) 276.15 K
 - b) -260 K
 - c) 260.15 K
4. In order to double the volume of gas, under constant pressure and at starting temperature of 27°C , it should be heated to a temperature of:
 - a) 200 K
 - b) 200°C
 - c) 600 K
5. If a constant force acts on a body, it will:
 - a) Stay at rest
 - b) Move at constant speed along straight line
 - c) Achieve constant acceleration
6. The SI unit for electric charge is:
 - a) Coulomb
 - b) Farad
 - c) Ampere
7. Electrostatic force between charges is:
 - a) Always attractive
 - b) Always repulsive
 - c) Attractive or repulsive depending on the sign of the charges

8. There are 5 moles of a gas in a vessel at 100°C . If the pressure of the gas amounts to 23.114 kPa and the gas constant is $R=8.314 \text{ J/Kmol}$, the volume of the gas equals:
- 0.06 dm^3
 - 0.67 m^3
 - 18.65 dm^3
9. A man walking 5.4 km/hour will travel the following distance each minute:
- 0.9 m
 - 90 m
 - 9 m
10. A bottle has a mass of 35.00 g when empty and 98.44 g when filled with water. When filled with another liquid, the mass is 88.78 g. What is the density of this other liquid?
- 0.8477 g/cm^3
 - 0.5378 g/cm^3
 - 0.6344 g/cm^3
11. Gamma rays are:
- Fast electrons
 - High energy photons
 - Slow neutrinos
12. Electric bulb of 110 W is connected in a circuit with 220 V voltage. The resistance of that bulb is:
- 220 Ω
 - 440 Ω
 - 50 Ω
13. A sharp image of an object is located 78 mm behind a 65 mm focal-length converging lens. The object distance from the lens is:
- 390 mm
 - 420 mm
 - 270 mm
14. When electrons are accelerated by 2450 V in an electron microscope, they will have wavelengths of:
- 8.113 nm
 - 0.622 nm
 - 0.811 nm
15. The radius of a hydrogen atom orbit where $n=2$ compared to the radius of a hydrogen atom orbit where $n=1$ is:
- Two times larger
 - Two times smaller
 - Four time larger
16. Is it possible to form an Ar_2 molecule?
- Not at all
 - Yes, of course

17. How many electrons there are in a molecule of water?
- 2
 - 8
 - 10
18. Electron mass is 9.1×10^{-31} kg. What is the proton mass?
- 9.1×10^{-31} kg
 - 1.7×10^{-27} kg
 - 1.7×10^{-31} kg.
19. The charge of the electron is 1.6×10^{-19} C. What is the charge of the proton?
- 1.6×10^{-19} C
 - 9.1×10^{-28} C
 - 3.2×10^{-16} C
20. How many neutrons are there in the nucleus of ${}^3_2\text{He}$?
- Two
 - One
 - Three
21. Nuclei of certain atoms emit α -particles. What is an α -particle, how would you designate it?
- ${}^4_2\text{He}$
 - ${}^4_2\text{He}^{4+}$
 - ${}^4_2\text{He}^{4++}$
22. What is the origin of the word ATOM?
- Little
 - Light (not heavy)
 - Indivisible
23. If you were locked in a vacuum with a piano, could you hear the music you play?
- Yes
 - No
24. If you were locked in a vacuum at night with a table lamp and a book. Could you read?
- Yes
 - No
25. If you drop two small balls 1 cm in diameter made of platinum and aluminium at the same instant (simultaneously) from the top of the Leaning Tower of Pisa, they will hit the ground
- At the same time
 - First the platinum ball an then aluminium one
 - First aluminium ball and then platinum one